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I. Introduction

The methyl radical, $CH₃$, is one of the most important building block fragment species in organic chemistry and has been found to play an important role in atmospheric,¹ interstellar² and combustion³ chemistry. Its (and its deuterated form CD_3) IR,^{4,5} UV,⁶⁻⁹ Raman^{10,11} and REMPI¹²⁻¹⁹ spectra have been thoroughly investigated to identify and characterize the ground state^{4,5,20} and electronically excited states.^{$6-9,11-13,15-19$} In terms of the electronically excited states the largest emphasis has been on studies of the lowest energy Rydberg states (CH₃**(Ry)), 3s ${}^{2}A_{1}$, ${}^{9-11,19,21}$ 3p₂ ${}^{2}A_{2}^{13,15-18,21,22}$ and ${}^{4}P_{z}$ ${}^{2}A_{2}^{13,14}$ with respect to spectroscopic properties^{9,11,13–19,21} and predissociation.^{10,11,14,16–19,21,22} The 3s and np_z ($n = 3,4$) states are typically accessed by one- (absorption) and two- (REMPI) photon excitations from the ground state $CH₃(X)$ A_2), respectively. Both Rydberg state spectra^{11,12,15} and photoelectron spectra²³ show a strong propensity for preservation of the vibrational states. The CH₃(X) (and CD₃(X)) radicals have been created by means of photolysis, $6,7,9-11,14,15,19-21,24-26$ dis-

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Photofragmentation and fragment analysis; Coriolis interactions in excited states of CH_{3} †

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Methyl radicals in their ground state $(CH_3(X))$ were created and excited by two- and one- color excitation schemes for CH₃Br and CH₃I, respectively, to record (2+1) REMPI spectra of CH₃ for resonant transitions to the Rydberg states CH₃**(np_z ²A₂); $n = 3$, 4. Various new and previously observed vibrational bands were identified and analyzed to gain energetic information for the Rydberg states. Particular emphasis was placed on analysis of the rotational structured spectra centered at 70 648 and 60 700 cm⁻¹, due to transitions from $v''_2 = 1$ to $v'_2 = 1$ and $v'_4 = 1$ for both Rydberg states, respectively. Vibrationally forbidden transitions between the v''_2 and v'_4 states, gain transition probabilities as a result of mixing of the ν_2 and ν_4 vibrational states due to Coriolis coupling between the two vibrational modes (intensity borrowing effect). As a result, the spectra are dramatically affected, both regarding line intensities and positions. This is the first direct evidence of a Coriolis interaction between two vibrational modes in Rydberg states of CH_3 (and in XH₃ molecules) based on simultaneous observation of spectra due to transitions to both interacting states. The analyses reveal close similarities between the Rydberg states and the ground state cation in terms of energy properties as well as the Coriolis interaction, as evident from comparison with recent work in relation to observation of $CH_{3}^{+}(X)$ in space (O. Berné et al., Nature, 2023, 621, 56–59). The effect of Coriolis interactions on predissociation of the Rydberg states/hence fragment products is discussed.

> charging,^{4,16} pyrolysis^{12,13,18} or reactions,^{13,17,27} prior to its studies or as a part of its formation analysis. Judging from observed spectra, $CH₃(X)$ is dominantly formed in the zero vibrational state and the v_2 (out-of-plane bending/OPLA) vibrational mode,4,11,13,15,17,20,24,26,27 while other vibrational modes are typically formed as v_1 (symmetric stretch) $> v_3$ (asymmetric stretch), ν_4 (deformation).^{17,26} Photolysis of CH₃I^{10,11,14,15,19,24,25} and CH_3Br ^{10,25} by one-photon excitations in the A-bands of the molecules are commonly used in the studies of $CH₃(X)$ and $CH₃^{**}(Ry)$ and relevant photodissociation studies reveal formation of CH₃(X) in different vibrational states $\left(CH_3(X; v''_1v''_2v''_3v''_4)\right)$ along with the halogen atoms in its ground $[I(3/2), Br(3/2)]$ or spin–orbit excited ($I^*(1/2)$, $Br^*(1/2)$) states, depending on the excitation wavelengths for $\text{CH}_{3}I^{24,28-36}$ and $\text{CH}_{3}Br^{26,37-41}$

> REMPI dynamic studies of CH₃X for X = I^{42} and Br,⁴³⁻⁴⁵ which involve two-photon resonant excitations to Rydberg states (CH_3X^{**}) in series that converge to the ground ionic states (CH_3X^+) , can also include one-photon non-resonant excitation to the repulsive valence states (CH_3X^*) corresponding to the A-band and three-photon excitations to superexcited Rydberg states $(CH_3X^{\#})$ above the ionization limit (Fig. 1a). The excited state intermediates CH_3X^* (for $X = I$, Br) and CH_3Br^{**} , have been found to dissociate to form $CH_3(X; v''_1v''_2v''_3v''_4) + X/X^*$, while $CH_3I^{\#}$ and $CH_3Br^{\#}$ predominantly form CH₃(X) + I^{**} (I^{**} = Rydberg state iodine atoms)⁴²

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Fig. 1 (a) Schematic figure for multiphoton excitations of CH₃X; X = I, Br and dissociation channels for formation of CH₃(X) and Rydberg states, CH₃**, via excitations to repulsive molecular valence states (CH₃X*), Rydberg states (CH₃X**) and superexcited states (CH₃X^{#)}) of CH₃X.^{42,44} (b) Excitation schemes relevant to the present studies of CH₃(X) (2+1) REMPI via Rydberg states, CH₃** for CH₃Br (two-color/pump and probe excitation scheme) and CH₃I (one-color excitation scheme).

and $CH_3^{**}(Ry) + Br/Br^*$,⁴⁴ respectively (Fig. 1a). The CH₃ Rydberg states formed in the three-photon excitations of CH_3Br via $CH_3Br^#$ depend on the photon energy as well as on the Rydberg electron symmetry (s, p, d) of the intermediate molecular Rydberg state, CH_3Br^{**} .⁴⁴

While the methyl cation, $\mathrm{CH_3}^+$ has long been considered to be an important precursor to organic chemistry in the universe $46-48$ it has only recently been discovered in a protoplanetary disk.49,50 Molecular IR emission detected by the James Webb space telescope from a strongly UV-irradiated medium was identified to be due to rovibrational transitions for the ${\nu_2}^+$ (OPLA) and ${\nu_4}^+$ (doubly degenerate in-plane bending) modes within CH $_3^{\mathrm{+}}$ (X) with strong Coriolis interaction between near-degenerate vibrational levels affecting the spectral structure significantly. Since the Rydberg states of $CH₃$ closely resemble the ground state of CH_3^+ (ref. 23) this discovery prompts consideration of whether Coriolis coupling between the corresponding vibrational modes ν_2 and ν_4 also may exist in CH_3 ^{**} Rydberg states.

Dynamic and spectroscopic experimental studies of fragment radicals such as $CH₃$ and its excited states require a selective formation step prior to the analysis of concern. Optimization of the species concentrations and the preferable conditions for the analysis step require simplicity in the initial step and may involve careful adjustment of a time delay between the two steps. In this paper we use a one-color multiphoton excitation scheme for CH3I and a two-color (pump and probe) excitation scheme for CH₃Br to create CH₃ $(X; v''_1v''_2v''_3v''_4)$ by absorption in the A-bands, followed by a $CH₃(X)(2+1)$ REMPI via $CH_3^{**}(4p_z^2A_2; v'_1v'_2v'_3v'_4)$ (for CH_3I and CH_3Br) and $CH_3^{**}(3p_z^2A_2; v'_1v'_2v'_3v'_4)$ (for CH_3Br) states (see Fig. 1b). The two-photon resonant excitation $\mathrm{CH_3}^{**}(4p_z^{-2}A_2) \leftarrow \leftarrow \mathrm{CH_3}(X)$ is in the same wavelength region as the one-photon excitation of $CH_3I^* \leftarrow CH_3I$, while that is not the case for the $CH_3Br^* \leftarrow CH_3Br$

excitation, which explains why the two different excitation schemes are needed. The data allowed comparison of the two different excitation schemes with respect to the quality of the spectral data. Furthermore, the data revealed new spectral observations and state characterizations relevant to the $CH_3^{**}(np_z^2A_2(n=3,4); v'_1v'_2v'_3v'_4) \leftarrow \leftarrow CH_3(X; v''_1v''_2v''_3v''_4)$ transitions, including the first observations of Coriolis couplings between the vibrational modes ν_2 and ν_4 in Rydberg states of $CH₃$, affecting the spectral structure dramatically and showing clear evidence of intensity borrowing effects. In addition, questions arise, what effects Coriolis interactions can have on fragmentation of $CH₃$ Rydberg states.

II. Experimental

Two multiphoton excitation schemes were applied to create the methyl radical in its ground state by photolysis of methyl halides, followed by $CH₃$ (2+1) REMPI via resonant excitations to Rydberg states (CH_3 ^{**}): one-color excitation of CH_3I in the two-photon energy region of 69 500-71 000 cm^{-1} (laser wavelengths 281.70–287.80 nm) and two-color/pump-and-probe excitation of $CH₃Br$ for initial pumping by a fixed wavelength within the A-band of the molecule, followed by probing in the two-photon energy regions of 59760-60800 cm^{-1} (laser wavelengths 328.90-334.7 nm) and 69 500-71 000 cm^{-1} (laser wavelengths 281.70–287.80 nm). The experimental apparatus and equipment parameters resembled those described in previous publications.⁵¹⁻⁵⁵

One-color excitation of CH₃I

Photolysis of CH₃I via its A-band absorption has emerged as a widely adopted and valuable method for generating the methyl radical (CH_3) . This approach is particularly convenient, as the laser wavelengths, 281.70–287.80 nm, necessary for the

two-photon excitation to Rydberg levels in the region of 69 500– 71 000 cm^{-1} falls within the A-band continuum. A CH₃I beam was created by a jet expansion of a gas mixture made by flowing argon over a $CH₃I$ liquid sample in a cooled trap, at a backing pressure of about 1 bar through a 500 µm pulsed nozzle into an ionization chamber. The pressure inside the ionization chamber was about 10 $^{\rm -6}$ mbar during experiments. The pulsed nozzle was typically kept open for about 140 ms and a laser excitation beam was typically fired 460 μ s after its opening. The excitation radiation was generated by a Nd:YAG laser (EKSPLA NL300 Series, 532 nm) pumped Coherent ScanMatePro dye laser followed by a frequency doubling with a BBO crystal. Dyes R610 and R590 were used. The laser beam was focused on the molecular beam by a 200 mm quartz focal lens between a repeller and extractor plates.

Two-color excitation of $CH₃Br$

Two laser systems consisting of pump lasers, dye lasers and frequency doubling units were used for the two-color excitation experiments. A $CH₃Br$ beam was generated by a jet expansion of a diluted gas mixture containing $CH₃Br$ in argon, typically with a CH_3Br : Ar ratio of about 1:1 to 1:5 for a backing pressure of 3 bar through a $500 \mu m$ pulsed nozzle into an ionization chamber. The pressure inside the ionization chamber was about 10^{-6} mbar during experiments. The pump laser excitation for photolysis of CH_3Br was generated by a Lambda Physik COMPex 205 Excimer laser pumped Lumonics Hyperdye 300 dye laser followed by a frequency doubling with a BBO crystal. Dyes C480 and C503 were used. Fixed wavelengths within the A-band of CH_3Br were chosen, typically 253 nm and 240 nm, prior to $CH₃(X)$ excitations to the 3p_z and 4p_z states, respectively. The probe laser system was the same as that used in the

one-color excitation scheme described above. Dyes R610, R590 and DCM were used. The probe laser pulses were delayed relative to the photolysis pulse on the ns time scale by a time delay unit (Quantum Pulse Generator 9520 series), a necessary disposition to optimize experimental conditions and ensure that an adequate density of photofragments had accumulated before the REMPI detection and/or to minimize ''fly-out'' of the photofragments. Typically, the delay was set to approximately 10 ns. The laser beams were focused on the molecular beam by 200 mm quartz focal lenses between a repeller and extractor plates.

For both excitation schemes ions were directed into a 70 cm long time-of-flight (TOF) tube and detected by a microchannel plate (MCP) detector to record the ion yield as a function of mass and laser radiation wavenumber. Signals were fed into a LeCroy WaveSurfer 44 MXs-A, 400 MHz storage oscilloscope. To prevent saturation effects and power broadening the laser power was minimized. Laser wavenumber calibration was based on observed iodine atomic (2+1) REMPI lines. The accuracy of the calibration was typically found to be about \pm 2.0 cm⁻¹ on the two-photon wavenumber scale.

III. Results and analysis

CH₃ formation and spectra

Fig. 2 shows CH₃ (2+1) REMPI spectra derived from $CH₃Br$ (two-color, pump and probe excitations; Fig. 2a) and $CH₃I$ (onecolor excitations; Fig. 2b; see also Fig. S1 in ESI† (ref. 56)) for the two-photon excitation regions of 59760-60800 cm^{-1} and 69 500-71 000 cm^{-1} , respectively (see also close up Fig. S2, S4 and S5 of spectra in ESI† (ref. 56)). Comparison with previous observations^{13,14,17,27} allows identification of

Fig. 2 Experimental (black, top) and calculated (blue, below) CH₃(X) (2+1) REMPI spectra for resonant excitations to the CH₃**(np_z²A₂) Rydberg states, $n = 3$ (a) and $n = 4$ (b), following one-photon ionization to form CH₃⁺. Experimental spectra are obtained by two-color/pump and probe excitation of CH₃Br (pump wavelength 253 nm) (a) and by one-color excitation of CH₃I (b). Vibrational transitions (0 $_{0}^{0}$, 1_{1}^{1} , 2_{1}^{1} and [2\41] $_{1}^{1}$) are indicated (see main text). Overlapping broad spectral features of CH₃⁺ signals following resonant excitations to Rydberg states of CH₃I are marked by * in (b). Rotational lines of calculated spectra are presented as Voigt shaped peaks with linewidths of 8 cm⁻¹ (a) and 3 cm⁻¹ (b).

spectra due to the transitions $CH_3^{**}(3p_z^2A_2; v_1'v_2'v_3'v_4') \leftarrow$ $CH_3(X; v_1''v_2''v_3''v_4'')$ for the vibrational transitions (0000) \leftarrow (0000) $(i.e. 0⁰₀; v₀ = 59972 cm⁻¹),¹³$ $(1000) \leftarrow \leftarrow (1000)$ $(1¹₁; v₀ =$ 59 881 cm⁻¹)¹⁷ and (0100) \leftarrow \leftarrow (0100) (2¹₁; v_0 = 60 700 cm⁻¹)¹³ and due to CH₃**($4p_z$ ²A₂; 0000) $\leftarrow \leftarrow$ CH₃(X; 0000) $(0_0^0; v_0 =$ 69 853 cm^{-1}).¹⁴ In addition to these spectra, a band peaking at 70 648 $\rm cm^{-1}$, about 791 $\rm cm^{-1}$ higher than the 0^{0}_{0} band centre peak of the CH₃^{**}(4p_z ²A₂) \leftarrow CH₃(X) transition at 69 857 cm^{-1} , is observed. This we assign to the corresponding 2_1^1 band based on the analogy with the 0_0^0 and 2_1^1 bands of the CH_3 ** $(3p_z^2A_2) \leftarrow \leftarrow \text{CH}_3(X)$ transitions, which are separated by about 728 cm^{-1} (see more detail below). 13 Furthermore, a band, prominently sticking out at 69 796 ${\rm cm^{-1}}$ about 61 ${\rm cm^{-1}}$ lower than the 0_0^0 band peak of the CH₃**(4p_z ²A₂) \leftarrow CH₃(X) transition at 69.857 cm^{-1} , is observed. This is assigned to the corresponding 1^1_1 band based on the analogy with the 0^0_0 and 1^1_1 bands of the CH₃^{**}(3p_z²A₂) \leftarrow CH₃(X) transitions, which are separated by about 77 $\rm cm^{-1}$ (see also Table 1). In addition to the observed $CH₃$ band spectra, broad spectral features are seen only in the spectra derived from $CH₃I$ (see Fig. 2b). Mass resolved REMPI spectra reveals the same spectral features for the CH_3I^+ and I^+ ions, allowing for identification of some spectra overlapping (underlying) the $CH₃$ bands (see Fig. 2b). These common broad spectral features indicate two-photon resonant excitations to CH₃I Rydberg states (CH₃I^{**}) followed by further excitation and branching processes to form the ions, CH_3^+ , CH_3I^+ and I^+ , thus being an indirect measure of twophoton absorptions due to the CH₃I^{**} \leftarrow CH₃I(X) transition. This is in accordance with our earlier observations based on analysis of ion and electron slice images recorded for lower energy REMPI excitations of CH₃I via CH₃I^{**},⁴² which revealed further one-photon excitation of CH_3I^{**} to superexcited state(s) $(\mathrm{CH_3I} ^{\#}),$

$$
CH_3I^{**} + h\nu \to CH_3I^{\#} \tag{1}
$$

followed by the branching (2) and photoexcitation (3) processes,

$$
CH_3I^{\#} \to CH_3(X) + I^{**}; \text{ dissociation} \qquad (2a)
$$

Table 1 Band origin, ν^0 (cm $^{-1}$) and assignments of spectra (see main text) due to transitions from the ground state to the $3p_z^2A_2''$ and $4p_z^2A_2''$ Rydberg states of $CH₃$. The numbers in parentheses are distances (in cm $^{-1}$) from the 0 $_0^0$ transitions

	$3p_z^2A_2''$		$4p_z^2A_2''$				
	Assignment Others work This work			Others work This work			
1_1^1	59 898 ^{a} (-76) 59 886 (-77)					$69796 (-61)$	
0 ₀ ⁰	59 881 ^b (-73) 59 974 ^{a} (0) 59 963 (0) 69 837 ^{b} (0) 69 857 (0) 59 954 b (0)			69 853 ^d (0)			
2^{1}_{1}	59972^{c} (0) 60688^a (714) 60711 (748) 60670^{b} (716)					70 648 (791)	
$\left[2\backslash4\right]_1^1$	$60\,700^c$ (728)	60 741 (778)				70 637 (780)	
	a^{a} Ref. 27. b^{b} Ref. 17. c^{c} Ref. 26. d^{d} Ref. 14.						

- $CH_3I^{\#} \rightarrow CH_3^+ + I^-$; ion-pair dissociation (2b)
	- $CH_3I^{\dagger} \rightarrow CH_3I^{\dagger} + e^{-}$; autoionization (2c)
- $I^{**} + h\nu \rightarrow I^+ + e^-$; photoionization (3a)
- $CH_3I^+ + h\nu \rightarrow CH_3^+ + I/I^*$ photodissociation (3b)

where I** are iodine atom Rydberg states. Thus, most likely, the ion formation channels (2b) and (3b) are responsible for the broad CH_3^+ signals, while channels (2c) and (3a) are responsible for the corresponding CH_3I^+ and I^+ signals, respectively (see ions highlighted in bold above). No overlapping (underlying) CH_3^+ spectral features, analogous to those observed in the one-color excitation scheme for CH₃I, were observed in the two-color excitation for CH_3Br . This is probably mainly due to the fact that negligible two-photon resonant excitations to CH_3Br^{**} , followed by further excitation and branching processes to form $\mathrm{CH_3}^+$, are to be expected in the region of the (2+1) REMPI probing of $CH₃(X)$.^{44,45,57,58} The different nature of the two-color excitation scheme for $CH₃Br$ (compared to the one-color excitation scheme for CH₃I), where CH₃ (X) is initially formed by pumping with a fixed wavelength chosen within the A-band of the parent molecule, followed by scanning at longer wavelengths after a time delay and optimization of conditions to detect $CH₃(X)$ (2+1) REMPI, could also be important. Considering also that the major channels following resonant excitations via Rydberg states within $CH₃I$ and CH₃Br involve further CH₃(X) formation along with I^{**} for CH_3I^{41} but CH_3 ** Rydberg states along with Br/Br* for CH_3Br^{44} (see Section I), the CH₃(X) spectra obtained by the use of CH₃Br are due to CH₃ $(X; v_1'v_2'v_3'v_4'')$ formed by photodissocation in the Aband of $CH₃Br$ only, while those derived from $CH₃I$ could also be affected by other CH₃(X) formation channels as well as the CH₃⁺ formation processes mentioned above. Further advantage of the two-color excitation scheme for CH₃Br over the one-color excitation scheme for CH₃I appears in the form of higher resolution spectra (sharper spectral peaks) obtainable in the former case, as a result of less power broadening. The $CH₃(X)$ spectra (Fig. 2) were further analyzed and simulated as described in the following section (see also ESI† (ref. 56)).

Spectral simulations

The CH₃ spectral structure, mentioned above, was simulated by using the PGOPHER program.⁵⁹ While the calculated spectra are based on the absorption cross sections for transitions between the states involved $(CH_3^{**}(Ry; v'_1v'_2v'_3v'_4) \leftarrow$ $CH_3(X; v''_1v''_2v''_3v''_4)$ only, the REMPI spectra depend on additional ionization of $CH_3^{**}(Ry; v'_1v'_2v'_3v'_4)$, which is not known in detail. This could cause some differences in line intensities while peak positions are not affected. Therefore, main emphasis was laid on matching peak positions (more than line intensities) of calculated and experimental spectra in the simulations. Energy levels $(E(N,K))$ of the states involved as a function of the rotational quantum number, N and the quantum number for the projections of the total angular momentum along the molecular symmetry axis (K) could be described, to a first approximation, as follows, $13,16,59-61$

$$
E(N,K) = T_{\rm v} + F(N,K) \tag{4a}
$$

$$
F(N,K) = BN(N+1) + (C - B)K^{2} - D_{N}N^{2}(N+1)^{2}
$$

$$
- D_{NK}N(N+1)K^{2} - D_{K}K^{4} - 2C\zeta lK
$$
(4b)

for all terms in cm^{-1} to give the transition wavenumber (ν) , for the energy difference between levels in the Rydberg states $(E(N',K'))$ and the ground state $(E(N'',K'')),$

$$
v = v^{0} + (F(N', K') - F(N'', K'')) \tag{4c}
$$

where T_v is the vibrational state energy, $F(N,K)$ is the rotational energy and v^0 is the band origin. The first five terms in eqn (4b) include standard spectroscopic terms for a symmetric top molecule (rotational constants, B, C, D_N , D_{NK} and D_K), the parameters given in ref. 4 and 62 for the ground state, the zero vibrational state and the vibrational mode ν_2 , respectively were used (see Table 2). The last term in eqn (4b) represents a Coriolis interaction strength between near-degenerate vibrational levels, used when appropriate (see below). ζ is the Coriolis coupling coefficient and $l (= \pm 1)$ is related to the vibronic angular momentum, defined based on molecular symmetry.⁵⁹ Relevant two-photon selection rules for rotational transitions between two electronic states are, 13

– for parallel transitions:

for
$$
K = 0
$$
, $\Delta N = 0, \pm 2$, $\Delta K = 0$
for $K \neq 0$, $\Delta N = 0, \pm 1, \pm 2$, $\Delta K = 0$ (5a)

– for perpendicular transitions:

$$
\Delta N = 0, \pm 1, \pm 2, \quad \Delta K = +1, \quad \Delta l = +1
$$

 $\Delta N = 0, \pm 1, \pm 2, \quad \Delta K = -1, \quad \Delta l = -1$ (5b)

The matrix elements due to Coriolis interactions between the two vibrational modes v_2 and v_4 in excited states for rotational levels of equal quantum numbers (N) for both vibrational states, as a function of N , K and l , for two-photon resonant excitation are expressed as,⁵⁹

$$
\langle v_2, N, K \pm 1, l \pm 1 | \hat{H}_{int} | v_4, N, K, l \rangle
$$

= $\sqrt{N(N+1) - K(K \pm 1)} \{ F_{ab} \pm F_{ab,c}(2K \pm 1) \}$ (6)

where \hat{H}_{int} is the interaction operator and F_{ab} and $F_{ab,c}$ are interaction strength parameters. The different signs (\pm) refer to shifts of the rotational levels in opposite directions, affecting the positions of corresponding spectral lines. In addition to line shifts, interaction can cause alterations in line intensities in the form of intensity borrowing, as a result of the state mixing.

(a) The 0_0^0 bands of the CH₃^{**}($np_z^2A_2$) $\leftarrow \leftarrow$ CH₃(X); *n* = 3,4, transitions.

The 0_0^0 bands for the CH₃^{**}(np_z ²A₂) $\leftarrow \text{CH}_3(X)$; $n = 3,4$, transitions both show a strong Q line in the middle and much weaker partly separated rotational lines of the O and P series on its short wavenumber side and the R and S series on its long wavenumber side (Fig. 2). Simulations of the rotational structure of the 0_0^0 bands for the parallel CH₃**(np_z ²A₂) \leftarrow CH₃(X); $n = 3,4$, transitions (Fig. 2) gave spectroscopic parameters for the zero vibrational states of the CH_3 ** $(3\text{p}_z$ ²A₂) and CH_3 **($4p_z$ A_2) Rydberg states in good agreement with values derived by Heinze et al^{16} and Black et al^{14} , respectively (Table 2; see also ESI† (ref. 56)).

Rotational line widths, which are believed to be affected by predissociation of the Rydberg states,^{14,16,21} were found to be larger for the 0^0_0 band of the 3p_z Rydberg state than of the 4p₂ state for comparable experimental conditions (see Fig. 2), with minimum FWHM of about 4.3 cm^{-1} for the 3p_z state but about 2.4 cm⁻¹ for the $4p_z$ state. This suggests that the $3p_z$ state is undergoing faster predissociation than the $4p_z$ state.

(b) The 2_1^1 and $[2\backslash 4]_1^1$ bands of the CH₃**($4p_z$ ²A₂) \leftarrow $CH₃(X)$ transition.

The spectrum in the two-photon excitation region of 70 520– 70 750 cm^{-1} resembles the structure of the 0^{0}_{0} bands by showing

<code>Table 2</code> Spectroscopic parameters for vibrational states of the ground- (top), 3p $_z$ 2 A $_2''$ and 4 p $_z$ 2 A $_2''$ states of CH $_3$ and for CH $_3^{\,+}$ (X) (bottom) (see main text). All values are in cm $^{-1}$ except for ζ , which is dimensionless. Detaild description of the parameter derivations and uncertainties for the $v_2'=1$ and $v_4'=1$ states are given in ESI⁵⁶

Parameters		Ref.	$T_{\rm v}$	В	C	$D_N \times \,10^4$	$D_{NK}\times 10^4$	$D_K \times 10^4$		F_{ab}	$F_{ab,c}$
$\tilde{X}^2A''_2$	$v'' = 0$	\boldsymbol{a}	$\mathbf{0}$	9.578	4.738	7.95	-14.46	6.873			
	$v''_2 = 1(a'_1)$	b	606.45	9.258	4.811	4.94	-7.06	2.57			
$3p_z^2A_2''$	$v' = 0$	с	59972	9.51	4.62	7.7	-13.6	6.3			
	$v'_2 = 1(a'_1)$	This work This work	59963 61318	9.47 9.18	4.84 4.2	9.4	-3.9	6.7			
	$v'_4 = 1(e'')$	This work	61348	9.96	4.6				-0.1	-8.9	0.02
$4p_z^2 A_2''$	$v' = 0$	d	69853	9.9	4.65	10.7	-21.2	$\mathbf{0}$			
	$v'_2 = 1(a'_1)$	This work This work	69857 71254	9.84 9.64	4.8 4.8	0.9 41.2	16.3	8.9			
	$v'_4 = 1(e'')$	This work	71243	9.78	4.51	-17.4			-0.14	-9.2	0.04
$\tilde{X}^{+1}A_1'$	$v^+ = 0$		79 357.3	9.361	4.615	7.26	-13.07	6.29			
	$v_2^+ = 1(a_2'')$	e	80 763.0	9.349	4.634	3.98	7.2	3.75			
	$v_4^+ = 1(e')$	\boldsymbol{e}	80 752.5	9.449	4.571	8.09	-13.7	6.16	-0.1	-8.9	0.02

^{*a*} Ref. 61. ^{*b*} Ref. 4. ^{*c*} Ref. 16. ^{*d*} Ref. 14. ^{*e*} Ref. 50.

Fig. 3 Experimental (black, top) and calculated (below) CH₃(X) (2+1) REMPI spectra for resonant excitations to the CH₃**(np_z²A₂) Rydberg states, n = 3 (a) and $n = 4$ (b), vibrational transitions 2_1^1 and [2\4]¹₁, following one-photon ionization to form CH₃⁺. Experimental spectra are obtained by two-color/ pump and probe excitation of CH₃Br (pump wavelengths, 253 nm (a) and 240 nm (b)). Rotational lines of calculated spectra are presented as sticks and as Voigt shaped peaks with linewidths of 8 cm $^{-1}$ (a) and 3 cm $^{-1}$ (b). Major rotational line series (Q lines) are indicated (see main text).

a broad peak in the centre (70648 cm^{-1}) with wings of rotational lines on both sides (Fig. 3b). This spectrum, initially assigned to the 2_1^1 band of the CH₃**($4p_z$ ²A₂) \leftarrow CH₃(X) parallel transition, as mentioned above, could not be simulated based on that assumption alone.⁵⁶ An assumption of an additional perpendicular transition to a state close in energy, interacting by Coriolis coupling, on the other hand, allowed the total structure to be simulated. This additional band we assign to the transition CH_3 ** $(4p_z \text{ }^2\text{A}_2; 0001) \leftarrow \leftarrow \text{CH}_3(X;$ 0100), *i.e.* a transition from the vibrational state $v''_2 = 1$ in the ground state to the vibrational state $v_4' = 1$ in the $4p_z^2A_2$ state, involving Coriolis coupling between the corresponding v_2 and ν_4 modes in the excited state. This we label as the [2\4]¹/₁ band, which overlaps the 2^1_1 band. While this transition is forbidden, according to the basic vibrational selection rule of ''no alterations in vibrational modes'' the state mixing, as a result of the Coriolis coupling, enhances the transition probability of the $[2\backslash4]^{1}_{1}$ transition and likewise diminish the transition probability for the 2^1_1 transition, *i.e.*, causes an intensity borrowing effect (see related work of ours in ref. 54 and references therein). The simulation of the combined rotational structure of the $2₁¹$ and $[2\backslash 4]_1^1$ bands for the CH₃^{**}(4p_z²A₂) $\leftarrow \leftarrow$ CH₃(X) transition is shown in Fig. 3b and the corresponding spectroscopic constants are listed in Table 2 (see also ESI† (ref. 56)). Fig. 3b shows how the total structure is built up of the two spectral bands 2^1 and $[2\backslash 4]_1^1$, separated by only about 11 $\mathrm{cm}^{-1}.$ The mixing of the two states, which is inversely proportional to the states/bands separation, is large enough to result in comparable total intensities of the 2^1 band $(I(2^1_1))$ and the $[2\backslash 4]^1_1$ band $(I([2\backslash 4]^1_1))$ $(I(2_1^1): I([2\setminus 4]_1^1) \sim 3:2$), corresponds to a large intensity borrowing effect. The Coriolis coupling between the two states acts in the form of level-to-level repulsion for levels of same K and N quantum numbers, inversely proportional to the level's energy difference, to cause line shifts depending on the Coriolis

coupling strength (see Fig. 4a, left). Thus, two ''interacting levels'' are ''pushed'' in opposite energy directions, as demonstrated in Fig. 4b (right). Furthermore, line intensities vary, depending on the interaction strength parameters $(F_{ab}$ and $F_{ab,c}$). Q lines dominate over much weaker O, P, R and S lines. The 2_1^1 band is mainly composed of Q lines ($\Delta N = 0$) for $\Delta K = 0$ (q lines), marked as ^qQ(N), while the [2\4]¹¹ band is mainly composed of Q lines ($\Delta N = 0$) for $\Delta K = -1$ (p lines) and $\Delta K = +1$ (r lines), marked as $PQ(N)$ and $PQ(N)$ lines, respectively (see Fig. 3b and 4b). A particularly prominent line series on the short wavenumber side of the center peak in the region of about 70 550–70 625 cm^{-1} turns out to be made, to a large extent, of ^rQ(*N*) rotational lines of the $[2\lambda 4]_1^1$ band for $N = 1-8$, $K'' = 0-7$ ($\Delta K =$ +1). Due to limited resolution of the spectrum the spectroscopic parameters of the excited state could not be determined with high precision. In the simulation procedure the Coriolis interaction parameters $(\zeta, F_{ab}$ and $F_{ab,c})$ were kept close to those determined for the corresponding v_2^+ vs. v_4^+ interaction in the ground state ion, $CH₃⁺(X)₅₀$ This is a justifiable assumption due to the structural similarities between $\text{CH}_3^+(X)$ and the Rydberg state $\text{CH}_3^{**}(4p_z^2A_2)$, which can be viewed as an ion core with an orbiting 4p Rydberg electron at a far distance, to a first approximation (i.e. $\text{[CH}_{3}^{+}\text{] }4\text{p}$) (see also ref. 23). Furthermore, the interaction parameters depend on the energy difference between the vibrational levels of the two vibrational modes, v_2 and v_4 , which is found to be similar (about 10 cm⁻¹ and 11 cm⁻¹ for CH₃⁺ and CH₃^{**}(4p_z²A₂), respectively).

Rotational linewidths are found to be comparable to that for the 0_0^0 band of the CH₃^{**}(4p_z²A₂) \leftarrow CH₃(X) transition, but smaller than for the 0_0^0 band of the CH₃^{**}(3p_z²A₂) $\leftarrow \leftarrow$ CH₃(X) transition for comparable experimental conditions, suggesting that the rate of predissociation of the $v'_2 = 1$ and $v'_4 = 1$ vibrational states are comparable to those of the zero vibrational state in CH_3 **($4p_z$ ²A₂), but less than for the zero vibrational state of CH_3 ** $(3p_z \text{ }^2A_2)$.

Fig. 4 Energy perturbations (a) and dominant two-photon excitations (b) in transitions to interacting v_2 and v_4 vibrational modes in CH₃**(np₂²A₂); n = 3, 4. (a), Left: Wavenumber differences between rotational lines ^qQ(N) and ^pQ(N) for K' = 0, ΔN = 1, derived from spectral simulations of the 2¹1 and [2\4]³ bands for the transition CH₃**(4p_z ²A₂) ← ← CH₃(X) (see main text and ref. 56). (a) Right: Schematic representation of the energy level shifts for N' = 6 and 5 of the v_2' and v_4' vibrational states for $K' = 0$, as a result of level-to-level repulsion between near-degenerate levels due to a Coriolis interaction between the vibrational modes, according to the line differences to the left. The arrows in the figure on left refer to the corresponding arrows in the figure on right. (b) Schematic representation of rotational energy levels for CH₃(X) and CH₃** (oblate symmetric top) and examples of transitions which dominate in transitions between levels of interacting vibrational states of the vibrational modes ν_2 and ν_4 for parallel (^qQ(N) lines; black arrows to left) and perpendicular (^pQ(N) and ^rQ(N) lines; red arrows to right) transitions.

(c) The 2_1^1 and $[2\lambda 4]_1^1$ bands of the CH₃**(3p_z 2 A₂) $\leftarrow \leftarrow$ $CH₃(X)$ transition.

The spectrum in the two-photon excitation region of about 60 560-60 800 cm⁻¹, peaking at about 60 700 cm⁻¹, which has been assigned to the 2_1^1 band of the CH₃**(3p_z²A₂) $\leftarrow \leftarrow$ CH₃(X)

transition, 13 does not show a resolved rotational structure (Fig. 3a). The spectral shape, however, could not be simulated based on that assumption alone,⁵⁶ while assuming a $[2\lambda 4]_1^1$ band close by and a Coriolis interaction between the $v'_2 = 1$ and $v_4' = 1$ vibrational states, analogous to the corresponding

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 CH_3 **($4p_z$ 2A_2) system described in the previous paragraph, allowed it to be simulated (Fig. 3a). The limited resolution of the spectrum, however, did neither allow determination of precise spectroscopic constant nor interaction parameters. Instead, the same Coriolis interaction parameters (ζ, F_{ab}) and $F_{ab,c}$) as those determined for the corresponding v_2^+ vs. v_4^+ interaction in $\operatorname{CH_3}^+(X)$ were used. 50 Furthermore, the rotational constants D_N , D_{NK} and D_K were set to zero to allow only three variables $(T_v, B \text{ and } C)$ for the fitting procedure. The simulation is shown in Fig. 3a and the spectroscopic constants are listed in Table 2 (see also ESI† (ref. 56)). The bands are now separated by about 30 cm^{-1} , compared to only about 11 cm^{-1} for the CH_3 **($4p_z$ ²A₂) state (see above). Due to the larger separation the mixing of the two vibrational states is less than for the CH_3 ^{**}($4p_z$ ²A₂) state, which appears in the form of larger total intensity ratio of the two bands of about $I(2_1^1): I([2\setminus 4]_1^1) \sim 5:1$, compared to the ratio of 3:2 for the CH_3 ** $(4\text{p}_z$ $A_2)$ state, corresponding to a less intensity borrowing effect. The 2^1_1 band is mainly made of Q lines for $\Delta K = 0$ (marked as ^qQ(N)), while the [2 $\{4\}^1_1$ band is mainly made of Q lines for $\Delta K = -1$ and $\Delta K =$ +1, marked as $PQ(N)$ and $PQ(N)$ lines, respectively as shown in Fig. 3a. Judging from the spectral simulation (see Fig. 3a) the lack of resolved rotational structure in the spectrum is due to high density of rotational lines with considerable linewidths.

While linewidths could not be determined from the simulation with acceptable accuracy, the overall structure could be simulated by assuming comparable linewidths as used for the simulation of the 0^0_0 band of the CH₃**(3p_z ²A₂) system. This might suggest that the rate of predissociation of the $v_2' = 1$ and $v_4' = 1$ vibrational states are comparable to that of the zero vibrational state within CH_3 ** $(3p_z \text{ }^2\text{A}_2)$.

IV. Discussions

The observation of Coriolis couplings between the two vibrational modes v_2 and v_4 within the Rydberg states CH₃**(np_z ²A₂; $n = 3, 4$), as evident from a combined spectral simulation of the 2_1^1 and $[2\backslash 4]_1^1$ bands, is an outstanding result of the data interpretation presented in the previous section. No such direct evidence of Coriolis interactions, based on simultaneous analysis of spectra due to transitions to both interacting vibrational modes, has been identified before, neither in $CH₃(X)$ nor $CH₃^{**}$. Coriolis interactions between vibrational modes either within the same electronic states or between different electronic states for XY_3 symmetric top molecules have long been known to exist. $63,64$ Numerous evidences for such interactions between bound and repulsive excited states, to cause predissociations, typically appearing in the form of spectral line broadening, have been found for $\mathrm{CH}_3, ^{14,16}$ NH $_3^{60}$ and SiH $_3.^{65}$ Spectral analyses relevant to either of the two interaction modes have revealed Coriolis interactions in the ground states of CH_3 ^{4,6,7} NH₃,⁶³ SiH₃⁶⁶ and BH₃⁶¹ or in the excited states of CH_3 ^{**} (ref. 67) and NH_3 ^{**}.⁶⁰ Thus, large anharmonicity observed for the ν_2 mode in the ground state of CH₃ in IR laser spectra⁴ allowed an evaluation of the Coriolis coefficient $(\zeta = 0.67697).$ ⁶⁷ While Hudgens *et al.* observed Coriolis interaction within the CH₃^{**}($n f^2 E$) (and CD₃^{**}($n f^2 E$)) state they could not find a significant Coriolis interaction within the CH_3 **(3p_z ²A₂)</sub> (and CD_3 **(3p_z ²A₂)) state, from rotational structure analysis of a $(2+1)$ REMPI spectrum for the $0⁰₀$ band for CD₃(X).¹³ The spectrum due to the CH₃**(3p_z ²A₂) \leftarrow $CH₃(X)$ transition, centered near 60 700 cm⁻¹, which has been observed in REMPI, $13,27$ has remained inexplicable and tentatively assigned to the $2₁¹$ transition only.^{13,27}

While a limited rotational structure of the spectrum centered near 60 700 cm^{-1} did not allow a detailed spectroscopic analysis, the corresponding spectrum $(2¹$ band) of the CH_3 ^{**}(4p_z ²A₂) \leftarrow CH₃(X) transition centered at 70 648 cm⁻¹ (Fig. 3b), which we observed, contained more structural information, promising for successful analysis. However, as mentioned before (see Section III), it could not be simulated by assuming it to be solely due to the 2^1 transition. Recent analysis of infrared spectra of the methyl cation and rotationally resolved photoelectron spectra of CH_3 ,⁵⁰ in connection with validating observations of radiation from $\mathrm{CH_3}^+$ in a protoplanetary disk,⁴⁹ turned out to be a key to solving the spectra analysis. Complex perturbations seen in the $\mathrm{CH_3}^+$ IR spectrum could be fully explained by assuming Coriolis interaction between the v_2^+ and v_4^+ vibrational modes within the ionic ground state and the relevant interaction parameters (ζ, F_{ab}) and $F_{ab,c}$) were derived. Since the Rydberg states CH $_3$ ** $(np_z\ ^2\mathrm{A}_2)$ can, to a first approximation, be considered to be made of the ground state ion core with orbiting Rydberg electrons at a far distance ($\left[\text{CH}_3^{\text{+}}(X)\right]$ *n*p), there is a reason to believe that similar Coriolis interactions are present in the Rydberg states. Furthermore, one might expect closer similarities between Rydberg states and the ground ionic state as the energy of the Rydberg states increases and approaches the ionization limit. Thus, by applying Coriolis interaction parameters close to those obtained for $\mathrm{CH_3}^+$ (see Table 2) and introducing simultaneous transitions to the two vibrational states $v'_2 = 1$ and $v'_4 = 1$ in CH₃**($4p_z$ ²A₂), close in energy, from the same original state (CH₃(X; 0100)), the spectrum in the region of $70\,520$ - 70750 cm^{-1} could be simulated for an energy difference between the two transitions of about 11 cm^{-1} (Fig. 3b and Table 2), which is comparable to the corresponding energy difference for CH₃⁺ of 10 cm⁻¹.⁵⁰ Analogously the spectrum in the lower energy region of 60 560–60 800 $\rm cm^{-1}$ (for $\rm CH_{3}^{**} (3p_{2}$ ${}^{2}A_{2}$)) could be simulated for larger energy difference between the $v'_2 = 1$ and $v'_4 = 1$ states of about 30 cm⁻¹. It should be emphasized, however, that even if the overall spectrum of the featureless structure of the CH₃**(3p_z $^{2}A_{2}$) spectrum could not be simulated without taking account of the Coriolis interaction, a single set of spectroscopic parameters could not be derived from the simulation.

Vibrational frequencies (in cm⁻¹) of the v_1 , v_2 and v_4 vibrational modes for the CH₃**(np_z ²A₂; n = 3,4) states, derived from the band origin (Table 1) and known vibrational frequencies of the ground state $(CH_3(X))$ are listed in Table 3, along with values derived by others for CH_3 ** $(3p_z \text{ }^2\text{A}_2)$ as well as values for CH₃(X) and CH₃⁺(X). As to be expected our ν_2 value

Table 3 Vibrational frequencies (in cm⁻¹) of the v_1 , v_2 and v_4 vibrational modes for the CH₃**(np_z ²A₂; n = 3,4) states, derived from the analyses of the band origin (Table 1) for known vibrational frequencies of the ground state (CH₃(X)) along with values derived by others for CH₃**(3p_z ²A₂), $CH_3(X)$ and $CH_3^+(X)$

State	Ref.	$T_{\rm v}$	v_1	v_{2}	v_{3}	v_4		
\tilde{X}^2A_2''	a	Ω	3004	607	3161	1398		
$3p_z^2A_2''$	This work	59963	2927	1355		1385		
	\boldsymbol{h}	59955	2931	1323	3085^c	(1428)		
$4p_{7}^{2}A_{2}''$	This work	69857	2943	1398		1387		
$\tilde{X}^{+1}A_1'$	d	79355	v_1 ⁺ 2943	v_2^+ 1405	v_3 ⁺ 3109	ν_4^+ 1395		
a Ref. 68. b Ref. 17. c Ref. 69. d Ref. 50.								

derived for the CH₃**(3p_z²A₂) state of v_2 = 1355 cm⁻¹ by taking account of the Coriolis interaction between the two modes differs significantly from the previous determination of v_2 = 1323 cm^{-1} obtained without considering any interaction.¹⁷ Even greater discrepancy is observed between our value of v_4 = 1385 cm⁻¹ and that determined by Zhang *et al.* of v_4 = 1428 cm^{-1} , which was estimated from data for CD₃ based on isotope shifts.¹⁷ Our values of $v_2 = 1398$ cm⁻¹ and $v_4 = 1387$ cm⁻¹ for the higher energy CH_3 ** $\text{(4p}_z\text{ }^2\text{A}_2)$ Rydberg states are larger than the corresponding values for the CH₃^{**}(3p_z²A₂) state and closer to those for CH₃⁺(X) of v_2 ⁺ = 1405 cm⁻¹ and v_4 ⁺ = 1395 cm⁻¹,¹⁷ as one might expect. Our value of $v_1 = 2943$ cm⁻¹ is identical to the corresponding value derived for v_1^+ of CH₃⁺(X).⁵⁰

Our minimum rotational linewidths (FWHM) of about 4.3 cm⁻¹ measured for the 0_0^0 band of the CH₃**(3p_z ²A₂) $\leftarrow \leftarrow$ $CH₃(X)$ transition (see above) is in good agreement with previous observations.16,28 This has been interpreted as being due to a combination of homogeneous and heterogeneous predissociation, of which the heterogeneous component is due to a Coriolis interaction with lower dissociation states to cause bound-to-free transition. 16,21 Our results, that the FWHM for the 0^0_0 and $2^1_1 + [2\lambda 4]^1_1$ bands of the CH₃^{**}(4p_z²A₂) \leftarrow CH₃(X) transition are significantly less (about 2.4 cm⁻¹; see above) than for CH₃**(3p_z ²A₂) $\leftarrow \leftarrow$ $CH₃(X)$ is also in agreement with other observations.¹⁴ This suggests that analogous predissociation channels for the $4p_z$ state are of less important than for the $3p_z$ state. The large linewidths, hence low resolution, of the 2_1^1 + $[2\backslash4]_1^1$ bands affect severely the quality of the parameters derived from the spectral simulations. Judging from linewidths of the 0^0_0 spectra for the $4p_z$ state 14 and lifetime measurements for the $3p_z$ states²² for CH₃ and CD₃ the linewidths of CD₃ spectra are likely to be smaller than for CH₃. Therefore, although the density of the rotational states is higher for CD_3 , recording and analysing the corresponding CD_3 ^{**}(4p_z) spectrum might be an alternative choice to improve the quality of the spectroscopic and interaction parameters.

V. Summary, conclusions and closing remarks

 $CH₃(X)$ radicals were created and resonant excited in multiphoton excitation of CH_3Br and CH_3I . A two-color pump and

probe excitation scheme was used in the case of $CH₃Br$, by pumping within the A-band of $CH₃Br$, followed by scanning the two-photon excitation regions of $59760-60800$ cm^{-1} and 69 500-71 000 cm⁻¹, which cover (2+1) REMPI of CH₃(X) for resonant excitations to the CH₃^{**}(3p_z²A₂) and CH₃^{**}(4p_z²A₂) vibrational Rydberg states, respectively. A one-color excitation scheme was used for $CH₃I$, scanning the two-photon excitation region of 69 500–71 000 cm^{-1} , to photoexcite CH₃I as well as to cover (2+1) REMPI of $CH₃(X)$ for resonant excitation to the CH_3 ^{**}(4p_z ²A₂) Rydberg state. While the one-color excitation scheme serves both the photolytic generation of $CH₃(X)$ and its probing (Fig. 1), it has the disadvantage that other $\mathrm{CH_3}^+$ REMPI signals, as a result of CH_3^+ formation channels following resonant excitations within CH₃I, overlapping those due to the $(2+1)$ REMPI of CH₃(X) (Fig. 2b). The pump and probe method for CH_3Br allowed better quality $CH_3(X)$ REMPI spectra, as a result of less power broadening, to be recorded.

New as well as previously observed REMPI spectra of $CH₃(X),^{13,14,17}$ were identified (Fig. 2) and analyzed (Tables 1 and 2).^{13,14,16,17,27} Detailed inspection of spectra centered at 69 796 cm^{-1} and 70 648 cm^{-1} revealed each of these to consist of closely separated bands as a result of transitions to interacting states. These were assigned to transitions from the same initial vibrational state, $v''_2 = 1$ of ground state CH₃(X) to the $v'_2 = 1$ (parallel transition, $2¹₁$) and $v'_4 = 1$ (perpendicular transition, $[2\backslash 4]_1^1$ vibrational states of the CH₃**(3p_z ²A₂) and CH_3 **(4 p_z 2A_2) states, respectively, affected by Coriolis couplings between the v_2 and v_4 vibrational modes. While the transitions from v_2 to v_4 ([2\4]¹₁) are vibrationally forbidden the transition probability is enhanced due to the mixing of the two vibrational modes due to the Coriolis coupling. The intensity borrowing effect is inversely proportional to the energy difference between the interacting states (see ref. 54 and references therein), appearing as larger total intensity ratio of the two band contributions $(I(2_1^1): I([2\setminus 4]_1^1))$ for the CH₃** $(4p_z^2A_2)$ state than for the CH₃^{**}(3p_z $^{2}A_{2}$) state. The combined spectral structure, due to the two transitions, is dominantly made of Q rotational lines ($\Delta N = 0$) for $\Delta K = 0$ (v_2 to v_2 /parallel transitions) and $\Delta K = \pm 1$ (ν_2 to ν_4 /perpendicular transitions). Spectral perturbations appear as irregularities in the line series due to the interaction in the form of level-to-level repulsions between levels of same rotational quantum numbers within the two vibrational states. Simulations of the combined 2^1 and $[2\lambda 4]^1$ bands allowed evaluation/estimation of spectroscopic parameters for both vibrational states, Coriolis interaction parameters and vibrational frequencies for both vibrational modes.

Judging from the analysis, the Rydberg states of concern closely resemble the ground state methyl cation in terms of the structure and vibrational and rotational properties, as well as the Coriolis interaction between the two vibrational modes, ν_2 and v_4 , as evidenced by comparison with a recent pioneering spectral analysis of $\mathrm{CH_3}^+(X)^{50}$ in relation to its recent observation in a protoplanetary disk.49 This is to be expected, since the Rydberg states CH_3 ** $(np_z{}^2\text{A}_2)$ can, to a first approximation, be considered to be made of the ground state ion cores with orbiting Rydberg electrons at far distances $([CH_3^+(X)]$ np).

Rotational linewidths, determined from the REMPI spectra, indicate a relatively short lifetime (τ) of the Rydberg states, CH₃**(np_z ²A₂) for $\tau(n = 3) < \tau(n = 4)$, due to predissociation of different rates. The predissociation is believed to be due to a combination of parallel and perpendicular transitions to repulsive states involving Coriolis interactions.^{16,21} It is, therefore a reason to believe that Coriolis interactions between excited states of $CH₃$, bound as well as repulsive, can influence the outcome of predissociation and photodissociation channels of CH3, in terms of fragmentation.

This paper demonstrates how the quality of data sampling relevant to fragment studies can depend on the method of its production by photolysis. Thus, two-color/pump and probe excitation is favored over one-color excitation in the case of spectral analysis of the fragment species of concern here $(CH₃(X))$. Energy properties of the fragment cation in its ground state proved to be an important guideline for predicting the corresponding energy properties of Rydberg states in a series converging to the ion. It allowed identification of spectral bands due to vibrationally forbidden transitions from the ground state to Rydberg vibrational states, with enhanced transition probabilities, hence intensity borrowing effect as a result of mixing with accessible vibrational states due to Coriolis coupling. Thus, the state interactions affect dramatically the spectra appearance. This is the first direct evidence of a Coriolis interaction between two vibrational modes in Rydberg states of $CH₃$ and in $XH₃$ type of molecules, based on simultaneous observation of spectra due to transitions to both interacting states. Furthermore, the interactions have implications for dissociation processes of Rydberg states in terms of fragment products. In general, we feel that this paper is important for shedding light on the effect of state interactions within molecules on energetic and fragmentation properties and believe that it can serve as a guideline for further work along those lines in fields such as astrochemistry and combustion.

Author contributions

Meng-Xu Jiang: data curation, formal analysis, funding acquisition, investigation, methodology, writing – review & editing. Agúst Kvaran: conceptualization, funding acquisition, investigation, methodology, project administration, supervision, validation, writing – original draft, writing – review & editing.

Data availability

Data supporting this article has been included as part of the ESI.† The code for PGOPHER (a program for rotational, vibrational and electronic spectra; C. M. Western, 2018, University of Bristol Research Data Repository) can be found at [https://](https://pgopher.chm.bris.ac.uk/) pgopher.chm.bris.ac.uk/, doi: [https://doi.org/10.5523/bris.3mq](https://doi.org/10.5523/bris.3mqfb4glgkr8a2rev7f73t300c) [fb4glgkr8a2rev7f73t300c](https://doi.org/10.5523/bris.3mqfb4glgkr8a2rev7f73t300c). The version of the code employed for this study is no. 10.1. Data, relevant to spectral simulations, including experimental and calculated spectra (bands 2^1 and $[2\backslash4]_1^1$; see main text) suitable for loading into PGOPHER are

available from University of Iceland at, [https://agust.hi.is/CH3/](https://agust.hi.is/CH3/CH3_4p_series.ovr) [CH3_4p_series.ovr](https://agust.hi.is/CH3/CH3_4p_series.ovr); https://agust.hi.is/CH3/CH3_4p_series.pgo, https://agust.hi.is/CH3/CH3_3p_series.ovr; [https://agust.hi.is/](https://agust.hi.is/CH3/CH3_3p_series.pgo) [CH3/CH3_3p_series.pgo](https://agust.hi.is/CH3/CH3_3p_series.pgo).

Conflicts of interest

There are no conflicts to declare.

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